## Claims

	[c1]	1. A process for removing SO $_2$ , NO, and NO $_2$ from a gas stream comprising
		the steps of
	Ü,	a. oxidizing at least a portion of NO in a gas stream to NO $_2$ with an
		oxidizing means, followed by
		b. scrubbing at least a portion of SO $_2$ , NO, and NO $_2$ from the gas
	÷	stream with a scrubbing solution
Start I		comprising ammonia, and
		having a pH between 6 and 8, and
		c. removing at least a portion of any ammonia aerosols generated from
		the scrubbing step from the gas stream with an aerosol removal means.
	[c2]	2. The process of claim 1, wherein said oxidizing means is an electrical
		discharge reactor.
	[c3]	3. The process of claim 2, wherein said electrical discharge reactor is a dielectric
		barrier discharge reactor.
Cont. Charles Street St	[c4]	4. The process of claim 3, further comprising the step of oxidizing at least a
		portion of the NO to HNO 3 with said dielectric barrier discharge reactor.
	[c5]	5. The process of claim 1, wherein said oxidizing step is adapted to result in a
		mole ratio of SO $_2$ to NO $_2$ of at least 2.5 to 1.
	[c6]	6. The process of claim 1, wherein said oxidizing step is adapted to result in a
		mole ratio of SO 2 to NO 2 of at least four to one.
	[c7]	7. The process of claim 1, said scrubbing solution
		comprising ammonia, ammonium sulfite, ammonium sulfate, and water,
		and
		having a pH between 6 and 8.
,	[c8]	8. The process of claim 1, wherein said aerosol removal means is a wet
		electrostatic precipitator.
	[c9]	

9. The process of claim 1, wherein said scrubbing step results in the formation

[c13]

of ammonium sulfate, the process further comprising the step of withdrawing ammonium sulfate from the scrubbing solution.

- [c10] 10.The process of claim 4, wherein said scrubbing step results in the formation of ammonium nitrate, the process further comprising the step of withdrawing ammonium nitrate from the scrubbing solution.
- [c11] 11.A process for removing SO  $_2$  , NO, NO  $_2$  , and Hg from a gas stream comprising the steps of
  - a. oxidizing at least a portion of the NO in a gas stream to NO  $_{\rm 2}$  , and at least a portion of the Hg in a gas stream to HgO, with an oxidizing means, followed by
  - b. scrubbing at least a portion of the SO  $_{\rm 2}$  , NO, and NO  $_{\rm 2}$  from the gas stream with a scrubbing solution

comprising ammonia, and

having a pH between 6 and 8, and

- c. removing at least a portion of any ammonia aerosols generated from the scrubbing step, and HgO, from the gas stream with an aerosol removal means.
- [c12] 12.The process of claim 11, wherein said oxidizing means is an electrical discharge reactor.
  - 13. The process of claim 12, wherein said electrical discharge reactor is a dielectric barrier discharge reactor.
- [c14] 14.The process of claim 11, wherein said aerosol removal means is a wet electrostatic precipitator.
- [c15] 15.The process of claim 11, said scrubbing solution comprising ammonia, ammonium sulfite, ammonium sulfate, and water, and having a pH between 6 and 8.
- [c16] 16.The process of claim 15, wherein said scrubbing step results in the formation of ammonium sulfate, the process further comprising the step of withdrawing ammonium sulfate from the scrubbing solution.

	comprising
	a. an oxidizing means for oxidizing at least a portion of the NO in a gas stream to NO $_{ m 2}$ , followed by
	b. a scrubber suitably adapted to scrub at least a portion of the SO $_2$ , NO,
	and NO 2 from the gas stream with a scrubbing solution
	comprising ammonia, and
	having a pH between 6 and 8, and
	c. an aerosol removal means for removing at least a portion of any
	ammonia aerosols generated by the scrubber from the gas stream.
[c18]	18.The apparatus of claim 17, wherein said oxidizing means is at least one
	electrical discharge reactor.
[c19]	19.The apparatus of claim 18, wherein said electrical discharge reactor is at
	least one dielectric barrier discharge reactor.
[c19]	20.The apparatus of claim 19, wherein said dielectric barrier discharge reactor is adapted to oxidize at least a portion of the NO to NO $_2$ and HNO $_3$ .
	21.The apparatus of claim 17, said scrubbing solution
	comprising ammonia, ammonium sulfite, ammonium sulfate, and water, and
[c21]	having a pH between 6 and 8.
[c22]	22. The apparatus of claim 17, wherein said aerosol removal means is at least
	one wet electrostatic precipitator.
[c23]	23. An apparatus for removing SO $_2$ , NO, NO $_2$ , and Hg from a gas stream
	comprising
	a. an oxidizing means for oxidizing at least a portion of the NO in a gas stream to NO $_2$ , and at least a portion of the Hg in a gas stream to HgO,
	followed by
	b. a scrubber suitably adapted to scrub at least a portion of the SO $_2$ , NO,
	and NO 2 from the gas stream with a scrubbing solution
	comprising ammonia, and
	having a pH between 6 and 8, and

17. An apparatus for removing SO  $_{2}$  , NO, and NO  $_{2}$  from a gas stream

[c17]

[c25]

[c26]

[c27]

[c28]

c. an aerosol removal means for removing at least a portion of any ammonia aerosols generated by the scrubber, and HgO, from the gas stream.

[c24] 24.An apparatus for removing SO  $_2$  , NO, and NO  $_2$  from a gas stream comprising

a. an NO oxidizer adapted to oxidize at least a portion of the NO in a gas stream to NO  $_{\rm 2}$  , followed by

b. a scrubber adapted to scrub at least a portion of the SO  $_{2}$  , NO, and NO  $_{2}$  from the gas stream with a scrubbing solution

comprising ammonia, and

having a pH between 6 and 8, and

c. an aerosol remover adapted to remove at least a portion of any ammonia aerosols generated by the scrubber from the gas stream.

25. The apparatus of claim 24, wherein said NO oxidizer is at least one electrical discharge reactor.

26. The apparatus of claim 25, wherein said electrical discharge reactor is at least one dielectric barrier discharge reactor.

27.The apparatus of claim 26, wherein said dielectric barrier discharge reactor is adapted to oxidize at least a portion of the NO to NO  $_2$  and HNO  $_3$  .

28.T h apparatus of claim 24, said scrubbing solution comprising ammonia, ammonium sulfite, ammonium sulfate, and water, and having a pH between 6 and 8.

[c29] 29.The apparatus of claim 24, wherein said aerosol remover is at least one wet electrostatic precipitator.